Intracellular Patch Electrochemistry: Regulation of Cytosolic Catecholamines in Chromaffin Cells

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Alterations in the cytosolic pool directly affect neurotransmitter synthesis and release and are suggested to be key factors in various neurodegenerative disorders. Although this cytosolic pool is the most metabolically active, it is miniscule compared with the amount of vesicular transmitter and has never been quantified separately. Here, we introduce intracellular patch electrochemistry (IPE), a technique that for the first time provides direct measurements of cytosolic oxidizable molecules in single mammalian cells. In amperometric mode, IPE detects total catechols, whereas in cyclic voltammetric mode, it preferentially measures catecholamines. In cultured chromaffin cells, the total cytosolic catechol concentration was 50–500 μM, of which ~10% were catecholamines. Reserpine, a vesicular monoamine transporter inhibitor, had no effect on the catecholamine pool but increased total catechols by fourfold to fivefold. Combined with pargyline, a monoamine oxidase inhibitor, reserpine increased catecholamine levels in the cytosol by approximately sixfold. Amphetamine induced a transient approximately fivefold accumulation of cytosolic catecholamines and a slow increase of total catechols. In cells incubated with 3,4-dihydroxy-L-phenylalanine (L-DOPA), catecholamines increased by ~2.5-fold and total catechols increased by approximately fourfold. Cytosolic catecholamines returned to control levels ~10 min after L-DOPA withdrawal, whereas total catechols remained approximately twofold elevated even after a 1.5 hr incubation in L-DOPA-free media. Our data indicate that cytosolic catecholamines are strictly maintained at a defined level, and drug-induced increases in their concentrations lead to the accumulation of other catecholamine derivatives, such as DOPAC and 3,4-dihydroxyphenylethylene glycol. These derivatives reside in the cytosol for hours after treatment and may be an underlying cause of drug-related cytotoxicity.

Key words: patch-amperometry; cyclic voltammetry; cytosolic catecholamine; chromaffin cells; reserpine; pargyline; amphetamine; L-DOPA

Introduction

The regulation of catecholamine metabolism in the cytosol has received much attention, because it has been shown that elevated catecholamines give rise to the formation of oxyradical species and catecholamine–protein adducts (Shen et al., 1996; Sulzer et al., 2000; Spencer et al., 2002). It has been suggested that abnormal catecholamine homeostasis may underlie the vulnerability of catecholaminergic neurons in various disorders including Parkinson’s disease (Stokes et al., 1999; Conway et al., 2001; Shastri, 2001) and drug abuse (Lotharius and O’Malley, 2000; Davidson et al., 2001; Larsen et al., 2002). In contrast, pharmacological treatments that alter available cytosolic catecholamine also regulate the number of catecholamine molecules released per exocytic event (Colliver et al., 2000; Sulzer and Pothos, 2000), suggesting that free catecholamine levels may play a role in synaptic plasticity. To understand mechanisms underlying dopamine (DA) neurotoxicity and the regulation of catecholamine secretion, it is crucial to understand the principles of catecholamine homeostasis in the cytosol.

Cytosolic catecholamines, which are believed to constitute only a small fraction of the total intracellular catecholamine content, have never been directly determined in mammalian cells. The development of electrochemical detection methods (Kissinger et al., 1973) and the introduction of the carbon fiber microelectrode (CFE) (Gonon et al., 1978; Ponchon et al., 1979) have provided a means to measure concentrations of a variety of oxidizable metabolites, including catecholamines. CFEs in amperometric (constant voltage) and cyclic voltammetric (CV; voltage ramps) modes have been extensively used to study the mechanisms of dopamine and norepinephrine exocytic release in different cell types (Wightman et al., 1991; Bruns and Jahn, 1995; Chen et al., 1995; Jaffe et al., 1998; Pothos et al., 1998). However, these methods provided little or no information about the regulation of the cytosolic pool of neurotransmitter. The first attempt to directly measure catecholamines in the cytoplasm was made by Chien et al. (1990) and Olefirovicz and Ewing (1990). Because of the large size of the giant dopamine neuron (GDN) of the pond snail, it was possible to insert the CFE directly in the neuronal cell.
body or perform a capillary electrophoresis of its intracellular metabolites. Another approach, which used an amperometric detection of catecholamines released from digitonin-permeabilized chromaffin cells, was used to study comparative changes in cytosolic catecholamines induced by different weak bases (Mundorf et al., 1999) but did not provide information about the absolute metabolite concentrations.

The recently introduced technique of cell-attached patch amperometry combines patch-clamp recordings of cell membrane capacitance and conductance with amperometric detection of released catecholamines using a CFE located inside a patch pipette. Patch amperometry has made simultaneous measurement of vesicle volume and the amount of catecholamine secreted from the same vesicle possible (Albillos et al., 1997). We combined patch-amperometric recording in the whole-cell configuration with ampermetric and CV detection of catecholamines within the cytosol of individual cells. Data obtained with this approach provide answers to longstanding questions about the mechanisms of cytosolic catecholamine regulation.

Materials and Methods

Chemicals were purchased from Sigma (Milwaukee, WI) unless otherwise stated. All stock solutions were prepared at 10 mM concentration. Reserpine and calcine-AM (Molecular Probes, Eugene, OR) were diluted in DMSO, amphetamine in EtOH, and 3,4-dihydroxy-L-phenylalanine (L-DOPA) in 0.1 N HCl. Corresponding solvents were added to control cultures. Animal protocols were approved by the Columbia University and Cornell University Institutional Animal Care and Use Committees.

Primary adrenal chromaffin cell cultures. Bovine chromaffin cells were prepared and cultured as described previously (Parsons et al., 1995). For rat-derived cultures, adrenal glands from 2- to 7-d-old Sprague Dawley rats were dissected in ice-cold HBSS. The capsule and cortex of adrenal glands were removed and the remaining medulla were minced. After several washes with HBSS, the tissue was incubated with Ca2+-free collagenase IA solution (250 U/ml) for 30 min at 30°C with stirring. The digested tissue was rinsed three times with HBSS and triturated gently in a solution containing 1% heat-inactivated bovine serum albumin and 0.02% DNase I. Disassociated cells were centrifuged at 3000 rpm to form a pellet and then resuspended in a culture medium comprised of DMEM, 10% fetal bovine serum, 50 U/ml penicillin, and 50 µg/ml streptomycin.

For amperometric and CV recordings from rat chromaffin cells, the bath saline A1 contained the following (in mM): 140 NaCl, 5 KCl, 2 MgCl2, 5 CaCl2, 10 glucose, 10 HEPES–NaOH, pH 7.2. Pipette extracellular saline B1 contained the following (in mM): 128 NaCl, 5 KCl, 2 MgCl2, 10 CaCl2, 10 HEPES–NaOH, pH 7.4. Pipette intracellular saline A3 contained the following (in mM): 145 CsGlu, 8 NaCl, 2 MgCl2, 2 ATP, 0.3 GTP, 10 glucose, 10 HEPES–NaOH, pH 7.3.

Square or sine wave voltage signals were applied to the CFE using an Axopatch 200B amplifier (Axon Instruments, Union City, CA) and a 3 M KCl solution in polyethylene tubing, was connected to the input of another Axopatch 200B amplifier was in contact with a bath saline (see Fig. 1A). Square or sine wave voltage signals were applied to this electrode to measure seal formation and patch disruption and monoxide changes in access resistance (Lindau and Neher, 1988; Thompson et al., 2001).

We found that in CV mode of detection, the output current was affected by suction pulses applied to the patch pipette. For CV measurements, the design of the electrode holder was therefore changed to isolate the compartment inside the polyethylene tubing from the patch pipette compartment (see Fig. 1B). Recording pipettes (borosilicate glass; wall thickness, 0.3 mm; outer diameter, 2 mm) were pulled with a micropipette puller (Model P-97; Sutter Instruments, Novato, CA) and fire polished before use. The typical pipette resistance in the bath was 1.5–5 MΩ.

Amperometry. For the amperometric mode of intracellular patch electrochemistry (IPE), the CFE potential was held at 700 mV. When the plasma membrane was ruptured by suction, substances diffusing from the cytosol into the patch pipette were observed as a slow wave of oxidation current. After baseline subtraction, the total area, the charge (Q), of this oxidation current wave was used to estimate cytosolic catecholamine concentration (C) using the following formula:

\[ C = \frac{Q}{n \cdot F \cdot V} \]

where \( n \) is the number of electrons transferred from a molecule to the CFE, which was assumed to equal two (Baur et al., 1988); \( F \) is Faraday’s constant, and \( V \) is the cytosolic volume.

Cyclic voltammetry. Data acquisition and analysis for IPE in CV mode were performed using a subroutine locally written in Igor Pro (WaveMetrics, Lake Oswego, OR). All programs used in the present study can be downloaded from our laboratory website (http://www.columbia.edu/~ds43/download.html). Voltage ramps, from −450 mV holding potential to 800 mV and then back to −450 mV over 10 msec (scan rate, 250 mV/msec), were applied at 90 msec intervals.

Volume calculations. Cytosolic volume and the volume inside the patch pipette that contributed to the dilution of molecules (see Results) were estimated from photographs taken before each recording (see Fig. 2A, B). The corresponding volumes were then calculated as (see Fig. 2B for parameters description):

\[ V_{cell} = \frac{4}{3} \cdot \pi \cdot R_{cell}^3 \]

and

\[ V_{pipette} = \frac{\pi}{2} \cdot (R_{pip}^2 + R_{pip}^2 + R_s \cdot R_e) \]

where \( h = h_{cell} \cdot \sqrt{2} \). Cultured bovine chromaffin cells were generally spherical, as shown in a three-dimensional reconstruction of a cell incubated for 15 min in solution A1 containing 100 µM calcien-AM and imaged using confocal microscopy (see Fig. 2C). However, some cells in rat chromaffin culture appeared somewhat flattened, in which case the volume estimate was reduced by a factor \( S \) to account for shape distortion. The coefficient \( S \) was 0.5, 0.75, or 1 depending on the shape of the cell under the microscope.

Calibration measurements. Amperometric calibration measurements for all concentrations of each compound were performed in triplicate with the same CFE and reference electrodes. CV calibrations obtained using different CFEs were normalized to the same electrode sensitivity (see Results). Current readings were taken 30–60 sec after the addition of compounds when the current was stable.

Statistical analysis. One-way ANOVA tests between different treatment groups were performed using Microsoft Excel program (Microsoft, Seattle, WA).
steps reflecting exocytosis of single vesicles. The capacitance is determined by applying a sine wave voltage between an additional electrode in the bath and the same reference electrode in the patch pipette, measuring the resulting current, and analyzing it with a lock-in amplifier (Albillos et al., 1997). Amperometric spikes and associated capacitance steps indicate exocytosis of individual vesicles (Fig. 1C, solid lines, arrows). The volume of each vesicle can be calculated from the vesicle capacitance, and the amount of catecholamine in the same vesicle is obtained by integrating the corresponding amperometric spike (Fig. 1C, top, dashed line), providing simultaneous measurements of quantal size, vesicle size, and vesicular catecholamine concentration (Albillos et al., 1997).

Cytosolic catecholamine concentrations can be determined using the same principles. When the patch was disrupted by a strong suction pulse, capacitance increased to an off-scale value as the whole-cell configuration was attained (Fig. 1C, asterisks). This allowed catecholamines to diffuse from the cytosol into the pipette, and a slow wave of oxidation current was measured by the CFE inside the patch pipette (Fig. 1C, top, solid line). In ~50% of the recordings, amperometric spikes similar to those observed for exocytic events were seen superimposed on this slow wave (Fig. 1C, arrowheads). These spikes presumably represented vesicles that diffused into the pipette and burst as they encountered the surface of the CFE. After appropriate baseline subtraction, the slow current wave was integrated (Fig. 1C, top, dashed line), and the amount of oxidizable molecules in the cytosol was estimated (see Materials and Methods). Superimposed amperometric events from bursting vesicles were subtracted to obtain the contribution solely from the cytosol (Fig. 1C, inset). The example shown in Figure 1C yielded a charge of ~611 pC, corresponding to 3.2 fmol catechols. The diameter of this cell calculated from its photograph was 20 μm, providing a cellular volume of 4.2 pl (Fig. 2A,B) and a cytosolic catechol concentration of ~700 μM.

Calibrations
The wave of oxidation current detected by amperometry represents all cytosolic oxidizable molecules. To determine the response of a CFE at 700 mV to various metabolites, the following compounds were tested: the catecholamines DA, norepinephrine (NE), and epinephrine (EP), their precursor L-DOPA, the catecholamine degradation products 3,4-dihydroxyphenylethylenglycol (DOPEG), 3,4-dihydroxyphenylacetate (DOPAC), homovanillic acid (HVA), and ascorbic acid (AA). Calibration curves for these compounds plotted in double logarithmic coordinates were all linear with slopes in the range of 0.6–0.73 and yielded similar amperometric currents at any given metabolite concentration (Fig. 2D, Table 1). Therefore, IPE in amperometric mode measures all oxidizable molecules in the cytosol, including all catechol derivatives and AA, with similar sensitivity.

IPE in CV mode
In contrast to constant voltage amperometry, CV differentiates between various metabolites on the basis of their redox properties. Catecholamines have specific oxidation–reduction profiles (voltammograms) with an oxidation peak at 350–450 mV for the voltage scan rate used here (Chow and von Ruden, 1995). We used IPE in CV mode to determine cytosolic catecholamines more exclusively. The current resulting from voltage ramps (see Materials and Methods) was recorded as a three-dimensional matrix of data (Fig. 3A). Current near the maximum of the catecholamine oxidation potential was sampled (Fig. 3C, indicated by
the red line in A) to determine the time corresponding to the maximal catecholamine concentration at the CFE. When present, baseline drifts on the sampled current trace were removed by subtracting the current sampled at 0 mV (Fig. 3A, blue line). Voltammograms of substances released from cells (Fig. 3B) were obtained by subtracting voltage–current sections of the three-dimensional data matrix measured at a time before patch rupture (an average of 10 consecutive sections at any time preceding the 30 sec time point) (Fig. 3F) from those at the maximum of catecholamine oxidation wave (an average of 10 sections at ~33 sec). Catecholamine concentrations were calculated from the height of their oxidation peaks (ΔI) on the voltammogram (Fig. 3B). Pseudo–three-dimensional reconstructions of changes in the voltammograms during the recordings were used to estimate the time course of catecholamine release from cells (Fig. 3D).

Calibrations
Substances listed in Table 1 were calibrated using CFEs in CV mode. In agreement with previous reports (Wiedemann et al., 1990; Pihel et al., 1996), CV was differentially sensitive to various catechol metabolites. At the same concentrations, DA, EP, and NE yielded 10–20 times higher oxidation peaks than DOPAC, DOPEG, and ascorbic acid in double logarithmic plots fit by linear functions. Calibration coefficients for these and other compounds are listed in Table 1.

had an identical dependence on the electrode surface (Fig. 4C). \( I_B \) was therefore used to normalize the responses of the individual electrodes as follows:

\[
\Delta I_{\text{norm}} = \Delta I \cdot \frac{I_{B}^{\text{std}}}{I_B}
\]

where \( \Delta I_{\text{norm}} \) is the normalized height of the catecholamine oxidation peak, and \( I_{B}^{\text{std}} \) is the reference background current.

To convert \( \Delta I_{\text{norm}} \) into a concentration, calibration measurements were performed over a wide range of DA (Fig. 4D) and other compounds levels (Table 1). Similar to amperometric calibrations, the data points for CV calibration curves were well fit by a linear function in double logarithmic coordinates. Metabolite concentration (C) may thus be obtained using the following formula:

\[
C = 10^\left(\frac{\Delta I_{\text{norm}}}{A} \cdot \frac{\pi}{10} \cdot \frac{1}{X} \right)
\]

where \( \Delta I_{\text{norm}} \) is in pA, C is in \( \mu \text{M} \), and A and B are the calibration coefficients from Table 1.

Dilution of molecules
As IPE in CV mode detects catecholamine concentration at the tip of the CFE, which is a few micrometers away from the cell, it was necessary to account for the dilution of molecules inside the patch pipette. We used two approaches to estimate the initial cytosolic catecholamine concentration. First, in the “geometric” approach, cellular volume and the volume of the patch pipette, from its tip to the tip of the CFE, were estimated from photographs taken before each recording (see Materials and Methods) (Fig. 2A, B), and the concentration of catecholamines at the CFE was multiplied by the calculated dilution factor, \( V_{\text{cell}} + V_{\text{pipette}} / V_{\text{CPE}} \). However, this method underestimated the true cytosolic catecholamine concentrations for two reasons: (1) there was a loss of molecules from the diffusion of catecholamines around the CFE into the shaft of the pipette, and (2) the actual free volume of the cytosol might be smaller than the total cell volume.

Table 1. Sensitivity of CFEs in amperometric and CV modes to different metabolic compounds

<table>
<thead>
<tr>
<th>Compound</th>
<th>Amperometry</th>
<th>CV</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>( A^0 )</td>
<td>( B^0 )</td>
</tr>
<tr>
<td>DA</td>
<td>0.86 ± 0.02</td>
<td>0.62 ± 0.04</td>
</tr>
<tr>
<td>NE</td>
<td>0.89 ± 0.02</td>
<td>0.61 ± 0.03</td>
</tr>
<tr>
<td>EP</td>
<td>0.85 ± 0.01</td>
<td>0.6 ± 0.02</td>
</tr>
<tr>
<td>I-DOPA</td>
<td>0.8 ± 0.01</td>
<td>0.67 ± 0.02</td>
</tr>
<tr>
<td>DOPEG</td>
<td>0.62 ± 0.02</td>
<td>0.73 ± 0.04</td>
</tr>
<tr>
<td>DOPAC</td>
<td>0.7 ± 0.02</td>
<td>0.68 ± 0.04</td>
</tr>
<tr>
<td>HVA</td>
<td>0.74 ± 0.01</td>
<td>0.64 ± 0.02</td>
</tr>
<tr>
<td>AA</td>
<td>0.69 ± 0.01</td>
<td>0.7 ± 0.02</td>
</tr>
<tr>
<td>CV</td>
<td>2.71 ± 0.03</td>
<td>0.72 ± 0.04</td>
</tr>
<tr>
<td>NE</td>
<td>2.39 ± 0.08</td>
<td>0.81 ± 0.07</td>
</tr>
<tr>
<td>EP</td>
<td>2.21 ± 0.28</td>
<td>0.8 ± 0.27</td>
</tr>
<tr>
<td>I-DOPA</td>
<td>1.64 ± 0.13</td>
<td>0.77 ± 0.09</td>
</tr>
<tr>
<td>DOPEG</td>
<td>1.64 ± 0.13</td>
<td>0.89 ± 0.1</td>
</tr>
<tr>
<td>DOPAC</td>
<td>1.31 ± 0.25</td>
<td>0.84 ± 0.16</td>
</tr>
<tr>
<td>HVA</td>
<td>1.54 ± 0.2</td>
<td>0.85 ± 0.15</td>
</tr>
<tr>
<td>AA</td>
<td>1.69 ± 0.11</td>
<td>0.5 ± 0.07</td>
</tr>
<tr>
<td></td>
<td>0.95 ± 0.15</td>
<td>0.79 ± 0.06</td>
</tr>
</tbody>
</table>

*a* Calibration coefficients were derived from the linear fit, \( \log(\Delta I) = A \cdot \log(C) + B \), where \( \Delta I \) is in picoamperes and \( C \) is in micrometers. Examples of the calibration curves are shown in Figures 2D and 4C. For CV calibrations, the reference background current, \( I_{B}^{\text{std}} \), was set at 30 nA. The coefficients are shown with the experimental errors.

*b* Current was calculated using the formula, \( \Delta I = 10^\left(A \cdot \log(\text{C}) + B\right) \), where \( \Delta I \) (pA) is an oxidative current at \( C \) (\( \mu \text{M} \)) concentration of a compound. For CV calibrations, \( \Delta I_{\text{norm}} \) values are presented (see Results).

*c* Ratio was calculated between \( \Delta I_{\text{at} 10 \mu \text{M}} \) for DA and other compounds.
because a portion of the cytosol is occupied by vesicles and other organelles.

The second approach used a random walk (RW) simulation (Berg, 1983; Sulzer and Pothos, 2000) to fit experimentally recorded sampled current traces and calculate the initial concentration of catecholamines inside the cell. The RW simulation, which was written in Excel (Microsoft), was based on two functions: diffusion of catecholamines from a spherical source of molecules through a small pore, and diffusion of molecules outside the cell from the pipette tip to the CFE. The first function was measured by Pusch and Neher (1988) and takes into account cell diameter, diffusion coefficient of catecholamines (\(D = 6.9 \times 10^{-6} \text{ cm}^2/\text{sec}\)) (Nicholson, 1999), and the access resistance of the pipette tip, which is a measure of the pore size at the pipette tip through which molecules diffuse from the cell into the pipette (Pusch and Neher, 1988). The second function describes the diffusion of catecholamines from a point source of molecules (the tip of the patch pipette) to the CFE. The initial intracellular concentration of catecholamines inside the cell was varied until the best fit of the experimental data was obtained. However, a weakness of this model is that it does not account for the restriction of diffusion by the pipette walls, thus, overestimating cytosolic catecholamine concentrations.

In the example of a single cell recording shown in Figure 3, the height of the oxidation peak determined from the voltammogram (Fig. 3B) taken at the maximum of the catecholamine oxidation wave was \(\Delta I = 377 \text{ pA}\), and the background current \(I_b\) was 18.8 nA. These currents then yielded \(\Delta I_{\text{norm}} = 602 \text{ pA}\) from Equation 1 and a catecholamine concentration at the CFE of 1.25 \(\mu\text{M}\) from Equation 2. The dilution factor calculated from cell and pipette geometric parameters was 2.9, yielding an estimate of 3.6 \(\mu\text{M}\) catecholamines in the cytosol. The RW best fit of the oxidation current wave (Fig. 3C, inset) yielded a 6.7 \(\mu\text{M}\) cytosolic catecholamine concentration. On average, the geometric method produced an estimate of cytosolic catecholamines 2.3 ± 0.6-fold lower than the RW simulation (mean ± SD; \(n = 9\)). Because the first method provided a lower estimate and the second method provided a higher estimate of the cytosolic levels of catecholamines, for the data presented below, an average of the two methods was approximated as follows: the catecholamine concentrations were calculated using the simpler geometric method and then multiplied by the factor of 1.65, derived from \((2.3 + 1)/2\).

**Figure 3.** Analysis of CV recordings from a single rat chromaffin cell. A. A representative recording shown as a three-dimensional matrix of data. Red and blue lines represent samplings of the current at 400 and 0 mV, respectively. B. Voltammogram of catecholamines released during the same recording. A subtraction was made between 10 consecutive averaged \(I–V\) profiles at 10 sec from those at 33 sec. Dashed line indicates \(\Delta I\), the height of DA oxidation peak. C. Current sampled at 400 mV oxidation potential. The asterisk indicates a time when the patch was disrupted by suction and a whole-cell configuration was attained. The inset shows a random walk fit (green) of the sampled current trace (red) using 15 \(\mu\text{M}\) access resistance, 8 \(\mu\text{M}\) cell radius, 60 \(\mu\text{m}\) distance between the CFE and the pipette tip, and 6.7 \(\mu\text{M}\) cytosolic catecholamine concentration. D. Pseudo-three-dimensional representation of approximate voltammograms of intracellular metabolites encountering the CFE during the recording, in which the intensity of color ("Planet Earth" color table) indicates the current at a given voltage and time. The algorithm used to calculate catecholamine concentration in this recording is discussed in Results.

**Subtraction of AA**

Although CV was ~50-fold more sensitive to catecholamines than to ascorbic acid (Table 1), it was necessary to account for the contribution of AA to the oxidation current, because it may reach high intracellular levels in chromaffin cells (Ingebritsen et al., 1980; Levine et al., 1985). On the basis of the markedly different shape of the AA voltammogram, an algorithm was developed to detect and subtract its contribution to the measured catecholamine voltammograms, as illustrated in Figure 4E. In this example, AA was added at 400 \(\mu\text{M}\) concentration followed by the addition of 5 \(\mu\text{M}\) EP. From \(-350\) to 0 mV, the ascorbate voltammogram showed a pronounced slope, whereas catecholamine voltammograms were essentially flat (Fig. 4E, bottom, left). The AA concentration was thus estimated from this slope on the voltammograms. The complete AA voltammogram was then approximated by three lines (bottom, middle, dashed line). This basic shape was assumed to be invariant, but the amplitude of its contribution was scaled in proportion to the measured slope between \(-350\) and 0 mV. The approximate scaled version of the AA voltammogram was then subtracted. The resulting difference closely reproduced the original voltammogram of EP (Fig. 4E, compare solid lines in left and right bottom panels).

Nearly one-half of rat chromaffin cells (90 of 190 cells) exhibited a slope between \(-350\) and 0 mV with the average \(\Delta I\) current of 1550 pA. Using AA calibration coefficients (Table 1), this corresponded to 700 ± 83 \(\mu\text{M}\) ascorbate (mean ± SEM; \(n = 90\)), although AA might not be the only factor that contributed to this oxidative current. For example, contributions from L-DOPA would also be partially subtracted because of the particular shape of its voltammogram (Fig. 4A). It should be noted also that in the amperometric mode of detection, the concentrations of catechols may be overestimated because of the presence of AA. These and other factors that may introduce uncertainties and cumulative errors in amperometric and CV measurements of catechol concentrations are summarized in Discussion.

![Image](image_url)
**Catecholamine levels in untreated chromaffin cells**

Cytosolic levels of oxidizable metabolites in untreated chromaffin cells measured by both amperometry and CV decreased with time in culture with a decay time constant of 3–4 d (Fig. 5A). Histograms of amperometrically detected compounds measured on day 1, days 2–4, and days 5–7 showed that the distributions from the younger cultures contained many cells with very high catechol concentrations (Fig. 5B). Distributions obtained on days 2–4 from bovine (Fig. 5B, middle) and rat (Fig. 5C, right) chromaffin cells had identical mean values of 290 ± 27 µM (n = 98) and 290 ± 21 µM (n = 97), respectively (mean ± SEM).

On day 1 after plating, the average concentration of oxidizable metabolites measured by amperometry was ~500 µM, whereas the catecholamine level estimated by CV was ~50 µM. This 10:1 ratio between amperometric and CV measurements remained almost unchanged over subsequent days (Fig. 5A). As discussed previously, amperometry is similarly sensitive to all tested metabolites, whereas CV is several-fold more sensitive to catecholamines DA, NE, and EP than to DOPA, DOPAC, or DOPEG (Table 1). Therefore, this is consistent with a 10:1 ratio between total catechol and catecholamine concentrations.

To minimize the influence of cell-to-cell variability, for additional experiments, we used cells that were cultured for 2–4 d when the relatively high concentrations of catechols were distributed with a minimal occurrence of outliers (Fig. 5C, right). The cytosolic concentration of catecholamines in rat adrenal cells at this time point was 21.5 ± 2 µM (mean ± SEM; n = 90), with >80% of the cells containing 5–35 µM catecholamines (Fig. 5C, left). Each group of recordings from drug-treated cells (see below) was matched with the control group from the same culture batch that was recorded on the same day.

Measurements performed in extracellular and low-calcium intracellular patch pipette solutions yielded the same catechol levels in both amperometric (saline A2 and A3) and CV (saline B1 and B2; see Materials and Methods for compositions) modes (Fig. 5D). No changes in cytosolic catecholamine homeostasis were observed during 1.5–2 hr of cell incubation in the recording solution (data not shown). No correlation was found between cytosolic volume and the concentrations of cytosolic catechols or catecholamines (data not shown).

**Perturbations of cytosolic catecholamine homeostasis in drug-treated cells**

Several pharmacological treatments have been suggested to alter intracellular catecholamine pools distribution in neurons and chromaffin cells, although the effects of these drugs on the cytosolic catecholamine homeostasis have never been measured directly. We used amphetamine, reserpine, pargyline, and L-DOPA to induce perturbations of catecholamine metabolism and monitored changes in the cytosolic catechol and catecholamine pools after drug treatments. Time courses of concentration changes were derived from measurements performed on individual cells incubated for the indicated periods.

**Amphetamine**

The changes in cytosolic catecholamine after exposure to amphetamine, methamphetamine, or methylenedioxymethamphetamine (ecstasy) have been a subject of intense conjecture,
attributable to both the mechanisms by which these drugs act and their neurotoxicity (for review, see Leviel, 2001). The actions of amphetamines include inhibition of mitochondrial monoamine oxidase (MAO) (Davidson et al., 2001), stimulation of catecholamine synthesis (Larsen et al., 2002), and induction of catecholamine reverse transport from the cytosol to the extracellular milieu by dopamine and norepinephrine uptake transporters (dopamine uptake transporter and norepinephrine uptake transporter) (Fischer and Cho, 1979; Sulzer et al., 1995; Sonders et al., 1997; Jones et al., 1998).

Extracellular amperometric recordings have confirmed that these drugs reduce the quantal size of transmitter release (Colliver et al., 2000); however, unlike amphetamine, reserpine and TBZ alone do not induce nonexocytic catecholamine release (Jones et al., 1998).

Using a similar treatment protocol as that for amphetamine, we detected no significant changes in cytosolic catecholamines during 60 min of chromaffin cell exposure to 1 μM reserpine (Fig. 7A, left) or 10 μM TBZ (data not shown). Catecholamine concentration in untreated cells and cells treated with reserpine for 20 min was 24.4 ± 6.9 μM (mean ± SEM; n = 9) and 20.0 ± 7.8 μM (n = 11), respectively. The amperometrically detected total cytosolic catechol levels increased approximately fourfold after reserpine treatment (Fig. 7A, right), indicating that catecholamines were rapidly metabolized, presumably to DOPAC and DOPEG, detected by amperometry but not by CV. In agreement with this, application of reserpine on cells pretreated with the MAO inhibitor pargyline (10 μM; 30 min) produced an increase in the levels of cytosolic catecholamines (Fig. 7A, left). Total catechol concentrations started to decrease slowly after ~30 min of reserpine application but were still >twofold higher than in control after 60–80 min (Fig. 7A, right).
L-DOPA

The effect of the catecholamine precursor L-DOPA on cytosolic catecholamines has been extensively studied because of its effects on neurotransmission and potential neurotoxicity from its administration to Parkinson’s disease patients (Fahn and Cohen, 1992). L-DOPA increases the quantal size in chromaffin cells by 70% (Pothos et al., 2002), suggesting that the availability of dopamine in the cytosol is a rate-limiting step in the process of catecholamine accumulation into vesicles.

Adrenal cells incubated with 100 μM L-DOPA for 1 hr displayed twofold to threefold higher cytosolic catecholamine levels (recordings made in the presence of L-DOPA) compared with untreated cells (Fig. 7B, left). After L-DOPA withdrawal from the extracellular media, catecholamine concentrations diminished rapidly and were close to controls after 10–30 min (5.6 ± 1.5 μM for control, n = 25; 7.1 ± 1.3 μM for l-DOPA, n = 6). Total cytosolic catechols were approximately fourfold elevated in cells preincubated with L-DOPA and remained approximately two-fold higher than control after 90 min of cell incubation in l-DOPA-free media (Fig. 7B, right).

**Discussion**

By combining patch amperometry and CV, we have developed intracellular patch electrochemistry, an approach that directly measures cytosolic oxidizable metabolites. Using this technique, we studied the regulation of catecholamine metabolism in the
cytosol of cultured rat and bovine adrenal chromaffin cells under
resting conditions and after treatment with a variety of agents
that have been previously suggested to redistribute intracellular
catecholamine pools.

Cytosolic catecholamine homeostasis

Although there are substantial differences between catechol-
amine exocytosis in chromaffin cells and neurons, they share
many of the features of catecholamine intracellular homeostasis,
including the presence of key enzymes and transporters (Bader et
al., 2002; Tischler, 2002). Catecholamines are synthesized in the
cytosol from tyrosine, which is converted to L-DOPA by tyrosine
hydroxylase (TH), the rate-limiting enzyme in catecholamine
synthesis, and then to DA by aromatic L-amino acid decarboxyl-
ase. After VMAT-mediated accumulation into secretory vesicles,
in chromaffin cells, DA is metabolized to NE by dopamine β-
hydroxylase, which requires AA as a cofactor. NE can return to
the cytosol, and, in some cells, is converted to EP that is again
taken up into the vesicles. A chromaffin cell may thus contain
DA, NE, and EP in the cytosol and primarily NE or EP in the
granules (Cooper et al., 1991). Negative feedback regulation of
TH by DA, and catecholamine degradation by the mitochondrial
MAO, are the two major mechanisms that maintain catechol-
amine concentrations in the cytosol at a low level. Measurements
of cytosolic pool of catecholamines in GDN of the pond snail
showed that ≥98% of intracellular catecholamine is bound (pres-
umably in vesicles) and not accessible to the cytoplasm (Chien et
al., 1990).

Using IPE in amperometric mode, we found that in chromaf-
fin cells, the cumulative concentration of cytosolic oxidizable
metabolites, including the catecholamines DA, NE, and EP, their
precursor L-DOPA and the degradation products DOPEG,
DOPAC, HVA, and AA ranged from 50 to 500 μM (Fig. 5). IPE in
CV mode, used to preferentially measure the catecholamines DA,
NE, and EP, showed that cytosolic levels of these metabolites were
2–50 μM (Fig. 5C, left), which was ~10% of the total cytosolic
catechol concentration. To date, this provides the only available
data on the absolute concentrations of catechols in the cytosol of
mammalian cells.

On the first day after plating, catecholamine and total catechol
levels were ~30–40% higher than in the subsequent 3 d (Fig. 5A),
accompanied by low exocytic activity of chromaffin cells during
the first day in culture (R. Staal, unpublished data). These data
implicate that culture preparation induced distortions of normal
intracellular catecholamine homeostasis, which led to increased
cytosolic levels and decreased vesicular catecholamine levels.

Unexpectedly, reserpine and TBZ, inhibitors of catechol-
amine uptake into the vesicles, had little effect on the cytosolic
concentrations of catecholamines measured by CV (Fig. 7A),
confirming the data from snail neurons (Chien et al., 1990). In
the presence of the MAO inhibitor pargyline, reserpine induced
an approximately sixfold increase of catecholamine levels in the
cytosol, suggesting that catecholamines redistributed from secre-
tory vesicles were rapidly degraded to DOPAC and DOPEG.
Consistent with this mechanism, the level of total catechols
greatly increased in the presence of reserpine (Fig. 7A). Surpris-
ingly, pargyline alone did not induce any detectable catechol-
amine alterations, presumably because of the exocytic quiescence
of chromaffin cells in culture and a slow rate of catecholamine
turnover in the cytosol.

Amphetamine treatment caused a substantial but short-lived
increase of cytosolic catecholamines, closely following the time
course of amphetamine-mediated catecholamine release ob-
served on neuronal preparations (Jones et al., 1998; Schmitz et al.,
2001). These data provide direct evidence for the displacement of
vesicular catecholamine into the cytosol by psychostimulants
proposed previously (Carlsson et al., 1963; Sulzer and Rayport,
1990). A slow accumulation of amperometrically detected cat-
echolamines during the following 90 min (Fig. 6) was consistent with
the formation of catecholamine degradation products, which was
delayed because of partial inhibition of MAO (Miller et al., 1980;
Davidson et al., 2001).

Chromaffin cells exposed to 100 μM L-DOPA showed a two-
fold to threefold increase in the concentration of cytosolic cate-
cholamines (Fig. 7B), in accordance with an increased quantal
size in chromaffin cells incubated with L-DOPA (Pothos et al.,
2002) and confirming the equilibrium between cytosolic and ve-
sicular catecholamine levels (Njus et al., 1986; Johnson, 1988).
Additionally, L-DOPA treatment induced a substantial enhance-
ment of catecholamine catabolism, which led to a fourfold to
fivefold elevation of the total catechol levels, in agreement with
previous reports (Ahlskog et al., 1996; Fisher et al., 2000). After
L-DOPA withdrawal, catecholamines returned to control levels
after ≈10 min, whereas total catechols remained approximately
twofold increased even after 1.5 hr of incubation in L-DOPA-free
saline. This difference in the rate of catecholamine and catechol
clearance from the cytosol, together with reserpine and amphet-
amine data, suggest that DOPAC transport from the cytosol to
the extracellular milieu is the rate-limiting step in cytosolic cate-
cholamine catabolism.

Together, our data indicate that in chromaffin cells, cellular
homeostasis maintains cytosolic catecholamines at a stable low
level under a wide range of changes in VMAT, MAO, and TH
activities. Interventions expected to increase cytosolic catechol-
amine concentration have relatively short-lasting effects. Exces-
sive amounts of DA, EP, and NE are rapidly converted to other
catecholamine derivatives, such as DOPAC and DOPEG. In con-
trast, elevated levels of catecholamine degradation products can
reside in the cytosol for hours after treatment and may contribute
to the quinone species formation and cellular oxynradical damage.

Methodological concerns

Using CV, we detected the presence of substantial amounts of
cytosolic metabolites other than catechols in ~50% of chromaff-
fin cells. In part, this can be explained by a higher susceptibility of
CFEs in the CV mode to various sources of experimental noise,
such as changes in pressure and salt composition after the patch
rupture. During patch formation, before intracellular metabo-
lites were able to diffuse into the patch pipette, these noises
were also observed in the CV but not in the amperometric mode of
IPE. Hence, a portion, if not most, of the noncatechol CV cur-
cents might be attributable to processes other than the oxidation
of cytosolic molecules and were not detected by amperometry.
However, if the real cytosolic concentrations of catechols were
lower than estimated by amperometry, the relative changes of the
total catechol pools induced by different drugs were greater than
discussed above.

Another concern is that a portion of the oxidation current
measured by IPE may represent vesicular rather than the cytoso-
lic pool of catecholamines. In this case, catechol levels will be
overestimated, especially in CV mode, which does not have the
time resolution to detect and subsequently subtract cate-
echolamines released from vesicles (Fig. 1C, arrowheads). How-
ever, our data with VMAT inhibitors argue against this possibility
because a depletion of vesicular catecholamine stores did not
affect cytosolic catecholamine concentrations measured by IPE
References


Lotharius J, O’Malley KL (2000) The parkinsonism-inducing drug in CV mode (Fig. 7A). In addition, the kinetics of amphetamine-induced changes in catecholamine level (Fig. 6A) can only be explained by the effects of this drug on the cytosolic but not on the vesicular pool of the hormone. Finally, previous studies on isolated chromaffin granules showed a low permeability of the vesicular membrane both to ions and the amine over a 24 hr period (Maron et al., 1983), implying that there should have been no catecholamine leakage from the vesicles during our recordings.

Several other factors can also lead to an underestimation or overestimation of the true cytosolic catechol concentrations measured by IPE. Overestimation of the cytosolic volume and the possible loss of molecules because of their diffusion into the shaft of the patch pipette (see Materials and Methods) would underestimate cytosolic metabolite levels. The assumption that DA represented most of the cytosolic catecholamines (because NE is synthesized inside the vesicles) may lead to an underestimation of catecholamine levels measured by CV, because DA yielded a higher CV oxidation current than NE and EP (Table 1). In addition, catecholamines can donate either two or four electrons per molecule under different conditions (Ciolkowski et al., 1994), therefore, introducing another uncertainty in our calculations (see Materials and Methods). It is difficult to ascertain whether any of these factors prevailed in our experiments or whether they negated each other, and the absolute concentrations of free cytosolic metabolites may be a few-fold higher or lower than estimated by IPE. Despite these concerns, a relative comparison of metabolite levels in cells treated with various agents is often of greater interest, and the suitability of IPE for this kind of measurement is supported by our data.

Overall, IPE provides a means to directly observe processes only hypothesized previously to be regulated by cytosolic catechol pools. Given the ongoing advances in electrochemical detection that now provide measurement of a broad variety of compounds, IPE may help to elucidate the regulation of diverse cytosolic metabolites, including classical neurotransmitters, nitric oxide (Villeneuve et al., 1998), reactive oxygen species (Navera et al., 1991; Privat et al., 1999), and ascorbate (Gonon et al., 1981; Cahill and Wightman, 1995). It may also prove useful for studies of transmembrane transport of metabolic compounds that can be coupled with the measurements of associated ionic currents.

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